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Create a written document that lists the issue, your reasoning, and your conclusion. SOLUTION Student answers will vary. NOTE TO INSTRUCTOR: See Appendix A of this Chapter 1 Solution

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a body with a mass m on an electronic scale, which is calibrated
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Solution: $p_{abs} = p_{atm} + p_{gage} = 101.3 \text{ kPa} + 20 \text{ in. H}_2\text{O}$
 248.8 Pa $1:0 \text{ in. H}_2\text{O} = 110.26 \text{ kPa}$ Ideal gas law: $\rho = \frac{p}{RT} =$
 $\frac{110.26 \text{ kPa}}{1 \text{ kgK} \cdot 2077 \text{ J} \cdot 1 \cdot 293.2 \text{ K}} = 1000 \text{ Pa}$ 1 kPa $\frac{\text{J}}{\text{Nm}}$ $\frac{\text{N}}{\text{Pam}^2}$ $\rho =$
 0.181 kg/m^3 b.) Situation: A sphere of 93 mm diameter contains an

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ideal gas. $T = 20\text{ C} = 293.2\text{K}$ Find: Calculate the density of argon at a vacuum pressure of 8.8 psi. Properties:

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