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Prentice Hall Physical Science: Concepts  
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(Laboratory manual), Annotated Teacher's  
Edition. Published by Prentice hall (2004)  
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Dissolving: Dissociation, Dispersion or  
Ionization Substances can dissolve in  
water by three ways: 1. dissociation 2.  
dispersion 3. ionization Dissociation is the

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**Action Answers**  
process by which an ionic compound separates into ions Ex: NaCl Dispersion is the process by which a compound breaks into small pieces that spread throughout the water Ex: sugar dissolves in water  
Ionization is the process by which neutral molecules gain or lose electrons (forming ions) Ex: hydrogen chloride gas & water

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THEORY Democritus (4th century B.C)

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... said the universe was made of invisible units called atoms In 1808, an Englishman named John Dalton proposed a theory: 1. every element is made of tiny unique particles called atoms that cannot be subdivided 2. atoms of the same element have the same ...



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Solids have a definite shape and volume  
because particles in a solid vibrate about a  
fixed point Adding/taking Energy out of a

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system When you add energy to a system the particles begin to move faster As they move faster they collide and create friction (Heat) As they continue we get a phase change Six Common Phase Changes phase change: is the reversible physical change that occurs when a substance changes from one state of matter to another Six

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common phase changes: melting, freezing

...

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In Action by ... Answers

Physical Science: Concepts in Action

Author(s): Frank et al. Publisher:

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physical science. An organized way of  
using evidence to learn about the natural  
the study of living things, including plants,  
animals, and mic. the study of the

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properties of matter and how matter  
changes.

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