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States that matter is composed of extremely small particles called atoms; atoms are invisible & indestructible;

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atoms of a given element are identical in size, mass, & chemical properties; atoms of a specific element are different from those of another element; different atoms combine in simple whole-number ratios to form compounds; in a chemical reaction, atoms are separated, combined, or rearranged.

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Modern Chemistry Midterm Study Guide 100 Questions

whenever two elements form more than one compound, the different masses of one element that combine with the same mass of the other element are in the ratio of small whole numbers. daltons atomic theory. 1) all matter is composed of atoms. 2)atoms of an element are identical in size mass and other properties.

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