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ionic crystal (a) has
mobile electrons in
the crystal c covalent
molecular crystal (b)
is hard, brittle, and
nonconducting a
metallic crystal (c)
typically has the
lowest melting point
of the four

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CHAPTER 10 REVIEW

States of Matter

SECTION 4 SHORT

ANSWER Answer the following questions in the space

provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches equilibrium, (a) the two opposing

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changes occur at equal rates. (b) there are no more phase changes. (c) one phase change predominates.

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molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical properties. Modern Chemistry Chapter 10 Review Answers States Of Matter

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