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(d) On the basis of your answers above, write the balanced chemical equation for the combustion reaction and determine the formula of the hydrocarbon. The partial pressures occur in the same proportions as the number of moles.

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The particle will be moving to the left when the velocity function is negative. So we look for:  $3 \cos t < 0$  for  $0 \leq t < 2\pi$

$v(t) = 3 \cos t$  for  $0 \leq t < 2\pi$

The particle will be moving to the left for  $t$  values between  $\pi/2$  and  $3\pi/2$ . b)  $6600$

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and what is expected for each, see “ The Free-Response Sections — Student Presentation ” in the AP Physics Course Description. 4. The scoring guidelines typically show numerical results using the value  $g = 9.8 \text{ m s}^2$ , but use of  $10 \text{ m s}^2$  is of course also acceptable. Solutions usually show numerical answers using both values

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1. Provides a solution based on a reasonable calculation, equation, or inequality from the answer given in part (b). 2. Clearly selects the next higher integer as the answer. Partially correct (P) if the response correctly completes component (1) listed above but not component (2). Incorrect (I) if the response does not meet the criteria for E ...

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